

Advanced Separations Chemistry FSC-505 Dr. Thomas A. Brettell Fall 2008

## Prerequisites: General Chemistry, Organic Chemistry, Instrumental Analysis, Forensic Chemistry

# **SYLLABUS**

# COURSE OBJECTIVES

1. Students will apply their knowledge of organic, inorganic, physical, and analytical chemistry in the understanding of analytical separations.

2. Students will apply their knowledge of chemistry to the concepts of sample preparation techniques.

3. Students will apply their knowledge of organic, inorganic, physical, and analytical chemistry in understanding the mechanisms of separation and applications in gas and liquid chromatography.

4. Students will build upon their knowledge of chemistry in understanding the concepts of gravimetric methods of such separation, such as precipitation.

## COURSE OUTCOMES

1. Students will be able to design a separations protocol, for the identification of an unknown organic substance.

2. Students will be able to design protocols for the gas chromatographic analysis of organic substances commonly found in forensic evidence.

3. Students will be able to design protocols for the liquid chromatographic analysis of organic substances commonly found in forensic evidence.

4. Students will be able to troubleshoot and identify the difficulties encountered in the separation of analytes common to forensic evidence.

5. Students will be familiar with the most recent developments in the field of separations reported in the analytical chemistry literature.

#### **REQUIRED MATERIALS FOR THE COURSE**

Available in the Campus Bookstore: Text:

Robert L. Grob and Eugene Barry. *Modern Practice of Gas Chromatography*, Fourth Edition. John Wiley & Sons, Inc., New Jersey: 2004.

Scientific hand calculator.

Instructor Information Lecture: Tues: 1:00 – 3:00 PM; Miller 136

Instructor's office: Oberkotter 6; Phone: 3495

Office Hours: Tues., Thurs. & Fri. 11 AM - 12 Noon

Email: Tabrette@cedarcrest.edu

## **GRADING SYSTEM**

The grade for this course is based on three take home exams (75%), and a written research paper (20%). The remaining 5% of the grade constitutes class participation and is determined by the instructor.

The final letter grade will be awarded according to the following scale:

| 93 - 100%  | Ā              | 80 - 82.9% | B     | 67 - 69.9% | $D^+$ |
|------------|----------------|------------|-------|------------|-------|
| 90-92.9%   | $A^{-}$        | 77 - 79.9% | $C^+$ | 60 - 66.9% | D     |
| 87 - 89.9% | $\mathbf{B}^+$ | 73 - 76.9% | С     | Below 60%  | F     |
| 83 - 86.9% | В              | 70 - 72.9% | C     |            |       |

## **POLICIES**

#### **Classroom Attendance**

Classroom attendance is **mandatory** in order to perform well in this class. If class is missed due to illness, sports event, or family emergence, the student is still responsible for any missed assignments and obtaining the lecture notes.

#### Written Research Paper

The student will be required to write a critical review of a topic (assigned by the instructor) in analytical separations related to forensic science. The paper will be a critical review of the recent literature (last 5 years of the Journal of Forensic Sciences and the Journal of Analytical Chemistry) on the topic chosen. The paper will not be less than 5 typewritten double-spaced pages or more than 10 typewritten double-spaced pages. All references within the Journal of Forensic Sciences and the Journal of Analytical Chemistry (according to JOFS guidelines) at the end of the paper. This research paper will account for 15% of the student's grade for the course.

#### **Obtaining Assistance**

Do not hesitate to seek assistance concerning class lectures, homework assignments, or grading. If the student can not make it to the instructor's scheduled office hours, she should make arrangements to meet with the instructor at an alternative time at both the student's and the instructor's convenience.

## Honor Philosophy

"The Cedar Crest College Honor Philosophy states that students should uphold community standards for academic and social behavior in order to preserve a learning environment dedicated to personal and academic excellence. Upholding community standards is a matter of personal integrity and honor. Individuals who accept the honor of membership in the Cedar Crest College community of scholars pledge to accept responsibility for their actions in all academic and social situations and for the effect their actions may have on other members of the College community."

The instructor expects each student to abide by the college's honor code. This honor code applies to all activities associated with this course. The student should realize that the honor code is an important aspect of the educational process at Cedar Crest College.

Cedar Crest College Faculty and Administration support the following statement concerning Classroom Protocol:

"Appropriate classroom behavior is implicit in the Cedar Crest College Honor Code. Such behavior is defined and guided by complete protection for the rights of all students and faculty to a courteous, respectful classroom environment. That environment is free from distractions such as late arrivals, early departures, inappropriate conversations, and any other behaviors that might disrupt instruction and/or compromise students' access to their Cedar Crest College education."

Please be sure to turn off all cell phones and pagers during class times.

## **Community Standards for Academic Conduct:**

"Academic integrity and ethics remain steadfast, withstanding technological change. Cedar Crest College academic standards therefore apply to all academic work, including but not limited to, handwritten or computer-generated documents, video or audio recordings, and telecommunications.

As a student at Cedar Crest College, each student shall:

- Only submit work which is his/her own.
- Adhere to the rules of acknowledging outside sources, as defined by the instructor, never plagiarizing or misrepresenting intellectual property.
- Neither seek nor receive aid from another student, converse with one another when inappropriate, nor use materials not authorized by the instructor.
- Follow the instructions of the professor in any academic situation or environment, including taking of examinations, laboratory procedures, the preparation of papers, properly and respectfully using College facilities and resources, including library and computing resources to ensure that

these resources may be effectively shared by all members of the College community.

- Abide by the Cedar Crest Computer Use Policy.
- If a student perceives a violation of the Academic Standards, he/she will go to their instructor.
- If you are unable to resolve the problem with the instructor, you should go to the chair of the department. If you need further assistance after the consultation with the instructor and the chair, you should see the Provost."

It is the instructor's policy to deal with violations of these Standards for Academic Conduct by awarding a grade of 0 for the assignment or examination in question.

## **Students with Learning Disabilities**

The instructor supports the Cedar Crest College policy regarding learning disabilities as follows:

"Students with documented disabilities who may need academic accommodations should discuss these needs with their professors during the first two weeks of class. Students with disabilities who wish to request accommodations should contact the Advising Center."

### **CHRONOLOGICAL PLAN FOR THE COURSE**

The schedule on the following page lists the topics, which constitute the lectures pertaining to this course. A summary of the testing schedule and content is provided. The schedule **may be** modified throughout the course, as needed.

### **Testing Schedule Summary**

| 2007 Date   | Test   | Material Included                            |
|-------------|--------|--|
| Tues. 9/23  | Exam 1 | Lectures 1 through 4 plus reading materials  |
| Tues. 10/28 | Exam 2 | Lectures 5 through 8 plus reading materials  |
| Tues. 12/2  | Exam 3 | Lectures 9 through 14 plus reading materials |

## FSC- 505: Advanced Separations Chemistry Fall 2008

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## **Lecture Topics Schedule**

|       | Lectu | re <u>Topic</u>   | Reading Assignment:        |  |
|-------|-------|---|----------------------------|--|
| 8/26  | 1     | Introduction to Analytical Separations<br>Simple Separation Procedures<br>Separations Involving Phase Changes | (Handout)                  |  |
| 9/2   | 2     | Separations Involving Gravimetric Methods:  | (Handout)                  |  |
| 9/9   | 3     | Separations Involving Extractions & Sample Preparation Techniques:  | Grob, Ch. 11               |  |
| 9/16  | 4     | Extractions (continued)   | Skoog, Ch 29               |  |
| 9/23  | 5     | Separations Involving Chromatography:<br>Gas Chromatography;<br>Exam I  | Grob, Ch. 1, 2             |  |
| 9/30  | 6     | Gas Chromatography (continued)  | Grob, Ch. 3, 4             |  |
| 10/7  | 7     | Gas Chromatography (continued)  | Grob, Ch. 6                |  |
| 10/14 |       | Fall Break!   |                            |  |
| 10/21 | 8     | Gas Chromatography/Mass Spectrometry  | Grob, Ch. 7                |  |
| 10/28 | 9     | Liquid Chromatography<br>Exam II  | Skoog, Ch. 28              |  |
| 11/4  | 10    | Liquid Chromatography (continued)   |                            |  |
| 11/11 | 11    | Liquid Chromatography/Mass Spectrometry   | (Handout)                  |  |
| 11/18 | 12    | EASTERN ANALYTICAL SYMPOSIUM  |                            |  |
| 11/25 | 13    | Separations Involving Electric Fields   | Skoog, Ch.30               |  |
| 12/2  | 14    | Separations Involving Miscellaneous Techniques<br>Forensic & Other Special Applications<br>Exam III           | (Handout),<br>Grob, Ch. 16 |  |

#### **BIBLIOGRAPHY:**

Ira S. Lurie and John D. Wittwer, Jr., <u>High Performance Liquid</u> <u>Chromatography in Forensic Chemistry</u>, (Marcell Dekker, Inc., 1983).

Ian Tebbett, <u>Gas Chromatography in Forensic Science</u>, (Ellis Horwood, 1992).

Harold M. McNair and James M. Miller, <u>Basic Gas Chromatography</u>, (John Wiley & Sons, Inc., 1997).

Lloyd R. Snyder, <u>Principles of Adsorption Chromatography</u>, (Marcel Dekker, Inc., 1968).

James M. Miller, <u>Chromatography: Concepts and Contrasts</u>, John Wiley & Sons, 1988).

Bruno Kolb and Leslie S. Ettre, <u>Static Headspace Gas Chromatography</u>, <u>Theory and Practice</u>, Second Edition, (John Wiley and Sons, Inc., 2006).

James S. Fritz and George H. Schenk, <u>Quantitative Analytical Chemistry</u>, Fourth Edition. (Allyn and Bacon, 1979).

I. M Koltoff, E. B. Sandell, E. J. Meehan, and Stanley Bruckenstein, <u>Quantitative Chemical Analysis</u>, Fourth Edition (The Macmillan Company, 1971).

Forensic Chemistry. Suzanne Bell. New Jersey: Pearson Prentice Hall, 2005.

Modern Practice of Gas Chromatography, Fourth Edition. Robert L. Grob and Eugene Barry. New Jersey: John Wiley & Sons, Inc., 2004.

Introduction to Modern Liquid Chromatography, Second Edition. L.R. Snyder. John Wiley & Sons, Inc., 1979.

<u>Fundamentals of Analytical Chemistry</u>, Seventh Edition. Douglas A. Skoog, F. James Holler, and Donald M. West. Philadelphia, PA: Saunders College Publishing, 1996.

<u>Principles of Instrumental Analysis</u>, Douglas A. Skoog, F. James Holler, and Timothy A. Nieman, Sixth Edition, Saunders College Publishing, 2006.

<u>Advances in Mass Spectrometry</u>, Chapter 2: "*LC/MS in Forensic Toxicology*", CRC Press, 2004.

Applied Electrospray Mass Spectrometry, B. Pramanik, Taylor & Francis, 2002.